



## Thermal-induced percolation in high-density polyethylene/carbon black composites

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### ABSTRACT

The networking of carbon black in high-density polyethylene melt has been traced by simultaneous electrical resistance ( $R$ ) and dynamic storage modulus ( $G'$ ) testing. Thermal-induced dynamic percolation was observed for both  $R$  and  $G'$  as a function of annealing time. The influence of temperature and filler volume fraction on percolation times to  $R$  and  $G'$  was investigated, and activation energies of dynamic percolation were determined. The percolation times were longer than the terminal relaxation time, and activation energies of dynamic percolation were significantly higher than those of viscous flow and molecular relaxation of the composites. A first order kinetics aggregation model was introduced into the classical percolation theory to account for the dynamic percolation. The contribution of filler networking was evaluated quantitatively.

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### 1. Introduction

In carbon black (CB) filled polymers, a conduction network is formed at CB volume fraction ( $\Phi$ ) above percolation threshold ( $\Phi_c$ ) [1] depending on the structure [2] and dispersion [3,4] of CB aggregates as well as the interaction between CB and the matrix [5–7]. The conductive composites are in thermodynamic nonequilibrium so that the formation of conduction network is highly dependent on temperature ( $T$ ) and time ( $t$ ) [8–11]. Thermal-treatment at elevated temperatures could accelerate the structural evolution of the percolation network [12,13]. In particular, annealing at temperatures above melting point ( $T_m$ ) of semicrystalline matrix causes CB aggregates to agglomerate to form a three-dimensional conduction network, which is accompanied by a drastic change of electrical resistance ( $R$ ), termed dynamic percolation [14].

Investigating the dynamic percolation is helpful for optimizing annealing conditions to improve positive temperature coefficient (PTC) performance of conductive composites based on semicrystalline polymers [15]. Dynamic light scattering has been utilized to determine the diffusion coefficient of colloidal particles [16–20] in polymer solutions in the dilute regimes.

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However, it is not applicable to the conductive composites highly filled with CB aggregates. On the other hand, CB has been used as a self-diagnosing probe to trace particle–polymer interaction involved in the relaxation and viscoelasticity of the matrix [21].

Dynamic rheology is sensitive to filler dispersion in polymers [22–26]. The presence of a temporary filler network is responsible for the low-frequency modulus plateau [27,28]. Empirical relationships have been established between dynamic storage modulus ( $G'$ ) measured at  $T > T_m$  and electrical resistivity ( $\rho$ ) measured at room temperature for CB-filled polymers [28–31].

Simultaneous measurement of conductive and rheological properties has been performed in CB-filled natural rubber [32], vulcanizates [33] or electrorheological suspensions [34] under shear actions. We have proposed a novel method for simultaneous measurement of  $R$  and  $G'$  for conductive polymer composites [35]. By using this method, we have investigated the simultaneous variations of  $R$  and  $G'$  for CB-filled semicrystalline polymers undergoing shear deformation [36], isothermal crystallization [37] or nonisothermal ramp [38]. In the present article, we focus our attention on the simultaneous measurement of  $R$  and  $G'$  as a function of  $t$  for high-density polyethylene (HDPE)/CB composites subjected to annealing at  $T > T_m$ . We attempt to clarify the influence of  $T$  and  $\Phi$  on CB networking and the resultant evolution of electrical conduction and dynamic rheological behaviors.

## 2. Experimental

### 2.1. Materials and sample preparation

HDPE (density  $0.954 \text{ g cm}^{-3}$ ,  $T_m = 130 \text{ }^\circ\text{C}$ , melt flow index  $0.09 \text{ g min}^{-1}$ ) was purchased from Yangzi Petrochem. Corp., China. CB (VX<sub>c</sub>-605, particle size  $25 \text{ nm}$ , density  $1.85 \text{ g cm}^{-3}$ , nitrogen absorption special surface  $63 \text{ m}^2 \text{ g}^{-1}$ , and dibutyl phthalate absorption  $1480 \text{ mm}^3 \text{ g}^{-1}$ ) was obtained from Shanghai Cabot Chem. Co., Ltd., China. Antioxidant (B215, relative molecular mass 647,  $T_m = 180\text{--}185 \text{ }^\circ\text{C}$ ) was a product from Ciba-Geigy Co., Japan. HDPE and CB were dried at  $70 \text{ }^\circ\text{C}$  for 30 h before use.

The HDPE/CB composites were prepared by mixing HDPE, CB and antioxidant on a two-roll mill at  $160 \pm 5 \text{ }^\circ\text{C}$  and 50 rpm for 10 min. Weight fraction of antioxidant was 2 wt% in the composites. The composites were prepared through compression molded at  $160 \pm 5 \text{ }^\circ\text{C}$  and 14.7 MPa for 10 min to form disc samples of 7.9 mm in diameter and 1.2 mm in thickness for simultaneous measurement. Rectangle sheets of 35 mm in length, 10 mm in width and 1.2 mm in thickness were also prepared and two pieces of copper net were mounted onto the opposite wide surfaces of the sheet to ensure good electrical contact with digital multimeter for resistance measurement at room temperature.

### 2.2. Room-temperature resistance measurement

Room temperature resistance measurement was performed at  $25 \text{ }^\circ\text{C}$ .  $R$  of the samples below  $1.2 \times 10^8 \Omega$  was measured using the two-probe method with an Escort-3146 digital multimeter.  $R$  above  $1.2 \times 10^8 \Omega$  was measured with a high-resistance meter (ZC36,  $10^{-14} \text{ A}$  and  $10^{17} \Omega$ ).

### 2.3. Dynamic rheology and electrical conduction simultaneous measurement

The setup for simultaneous measurement of  $R$  and  $G'$  was described elsewhere [35]. Rheological measurement was performed on an advanced rheometric expansion system (ARES, Rheometrics Co., USA) in dynamic time sweep mode under 1% strain with a frequency ( $\omega$ ) of  $1 \text{ rad s}^{-1}$ . Simultaneously,  $R$  was measured with an automatic measurement system where a 7150 plus digital multimeter (Schlumberger) was installed. The applied direct current voltage was 1 V. Experiments with different strain amplitudes showed that the influence of 1% strain on  $R$  was negligible [39]. Dynamic frequency sweep was conducted from  $10^2 \text{ rad s}^{-1}$  to  $0.0158 \text{ rad s}^{-1}$  under 1–5% strain in the linearity region.

## 3. Results and discussion

### 3.1. Percolation at different temperatures

Fig. 1 shows  $\rho$  as a function of  $\Phi$  for the HDPE/CB composites at room temperature. Also shown are the data measured at  $160 \text{ }^\circ\text{C}$ ,  $180 \text{ }^\circ\text{C}$ ,  $200 \text{ }^\circ\text{C}$  and  $220 \text{ }^\circ\text{C}$  for the composites being annealed at the corresponding temperatures for 3 h. The data at room temperature exhibits the typical percolation transition.  $\rho$  drops dramatically by more than 10 orders of magnitude with increasing  $\Phi$  above  $\Phi_C = 0.05$ . The curves at elevated temperatures show a similar tendency except for a slight increase in  $\Phi_C$  to 0.07. At  $\Phi < \Phi_C$ ,  $\rho$  at elevated temperatures is over four orders lower than that at room temperature, which might be ascribed to the thermal-excited conduction in the matrix. On the other hand, at  $\Phi \gg \Phi_C$ ,  $\rho$  at elevated temperatures is higher than that at room temperature, which is attributed to the dilution effect resulting from the matrix melting.

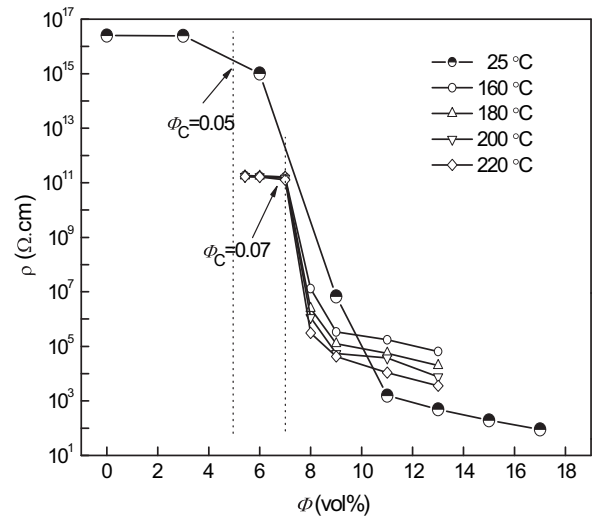


Fig. 1. Volume resistivity ( $\rho$ ) as a function of CB volume fraction ( $\Phi$ ) for HDPE/CB composites at room temperature and at elevated temperatures after being annealed for 3 h.

### 3.2. Dynamic percolation

Fig. 2 presents  $R$  and  $G'$  as a function of  $t$  for the composite ( $\Phi = 9 \text{ vol}\%$ ) during annealing at four different temperatures above  $T_m$ . At the earlier stage of annealing,  $R$  and  $G'$  change slightly with  $t$ . After critical times denoted as  $t_{pR}$  and  $t_{pG}$  on the figure,  $R$  decreases while  $G'$  increases markedly with increasing  $t$ , appearing as “dynamic conduction (DC) percolation” and “dynamic modulus (DM) percolation”.

Excess interfacial energy of conductive composites is considerably high. CB aggregates in the melt tend to agglomerate in order to reduce excess interfacial energy [5,6], which is accelerated due to annealing at elevated temperatures [40]. The driving force for the agglomeration might be the strong dispersive interaction between CB and the matrix as well as the depletion interaction between adjacent CB aggregates [41]. Wu et al. [21] investigated the DC percolation of CB-filled polymers and ascribed the sharp  $R$  decay at  $t > t_{pR}$  to the formation of CB network throughout the matrix. In the work of this paper,  $R$  and  $G'$  were recorded simultaneously during

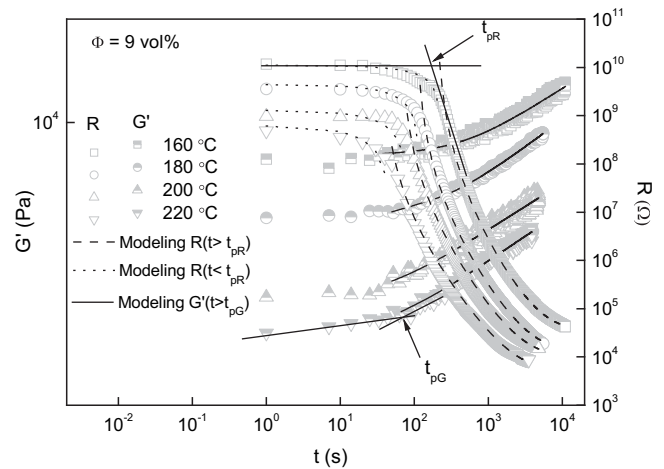


Fig. 2. Storage modulus ( $G'$ , half-filled symbols) and resistance ( $R$ , hollow symbols) as a function of time for the composite with  $\Phi = 9 \text{ vol}\%$  at four temperatures. The curves are fitted according to the modified percolation model.

annealing. When the composites were heated to a temperature above  $T_m$ , the conduction network preexisting in the semi-crystalline state was partially broken down. The  $R$  evolution in Fig. 2 thus reflects the reformation of conduction pathways in the melt.

Addition of CB into polymers leads to dramatic changes in viscoelasticity of the melt. The modulus increment due to filling can be mainly ascribed to direct bridging of single chains adsorbed on different aggregates, entanglement between chains adsorbed on separate aggregates, and entanglement of non-adsorbed bulk chains [22,27,42]. Three types of network, a temporary filler network, a temporary polymer network formed by entanglement, and a combined filler-polymer network, have been proposed for explaining the rheological behavior of filled polymers [43]. Incorporation of CB into polymers generally results in a reduction in mobility of polymer chains between CB aggregates [44,45]. At given temperatures, the marked  $G'$  increment with  $t$  at  $t > t_{pG}$  can be ascribed to the gradual formation of the combined network. Similar rheological evolution has been observed in nanotube-filled polymers [46–48]. As shown in Fig. 3,  $t_{pG}$  is generally smaller than  $t_{pR}$ , suggesting that the combined network responsible for the DM percolation is easier to form in comparison with the CB network responsible for the DC percolation [48].

### 3.3. Activation energies of dynamic percolation

Fig. 3 gives the critical times  $t_{pR}$  and  $t_{pG}$  as a function of reciprocal temperature ( $1/T$ ). At given temperatures, both  $t_{pR}$  and  $t_{pG}$  decrease significantly with increasing  $\Phi$ . Activation energies of the DC percolation and the DM percolation,  $\Delta E_R$  and  $\Delta E_G$ , were determined from the slopes of  $t_{pR}$  and  $t_{pG}$  against  $1/T$  and are shown as a function of  $\Phi$  in Fig. 4. Both  $\Delta E_R$  and  $\Delta E_G$  are nearly independent of  $\Phi$  at  $\Phi = 8$ –11 vol%.

It is reported that the aggregation of CB in polymer melts has a close relationship with polymer dynamics [21,49]. To compare the activation energies of dynamic percolation and viscoelasticity, we measured complex viscosity ( $\eta^*$ ) as a function of  $\omega$  for the composites at 160 °C, 180 °C, 200 °C, and 220 °C. Fig. 5a shows master curves of  $\eta^*$  against  $\omega$  for the HDPE/CB composites at 160 °C. Fig. 5b shows the horizontal and the vertical shift factors,  $\alpha_\eta$  and  $\alpha_\omega$ , used for constructing the master curves, as a function of  $1/T$ .  $\eta^*$  increases with increasing  $\Phi$  due to the reinforcement effect. The Arrhenius plots of  $\alpha_\eta$  and  $\alpha_\omega$  against  $1/T$  in Fig. 5b were used to determine activation energies of viscosity and frequency,  $\Delta E_\eta$  and  $\Delta E_\omega$ , and the values are shown in Fig. 4.

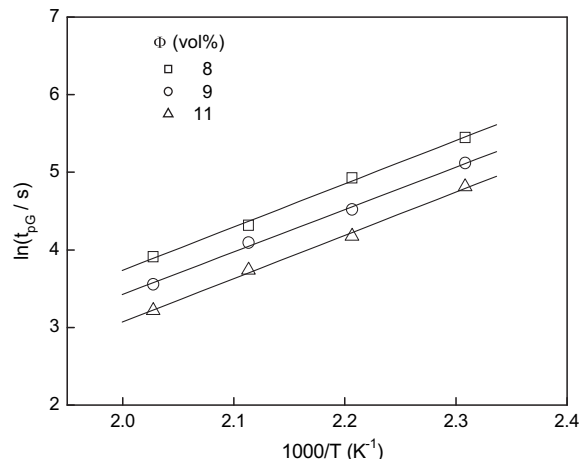
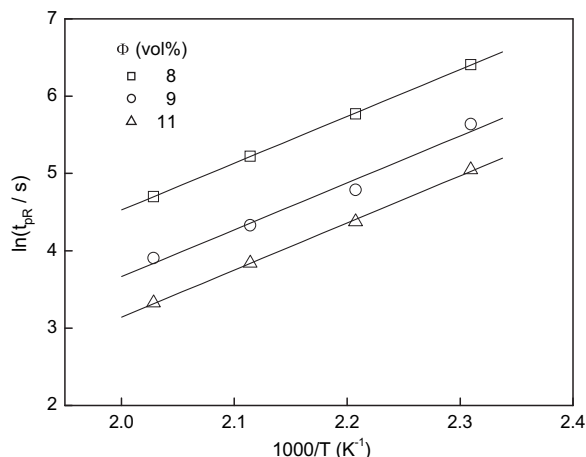


Fig. 3. Arrhenius plots of resistance and modulus percolation times ( $t_{pR}$  and  $t_{pG}$ ) versus reciprocal temperature ( $1/T$ ) for the composites.

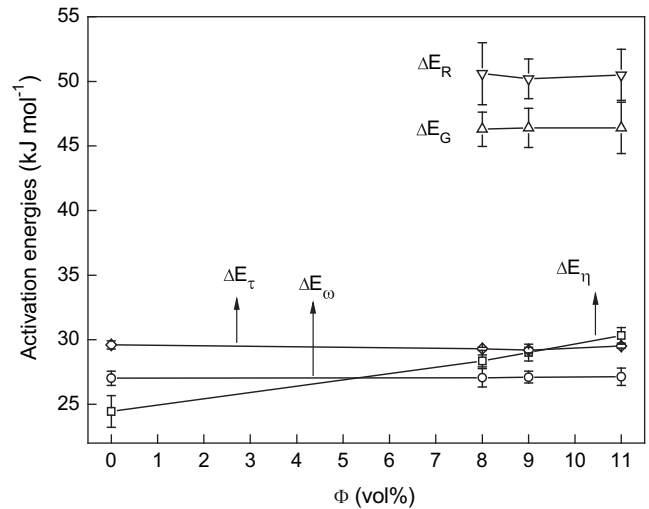
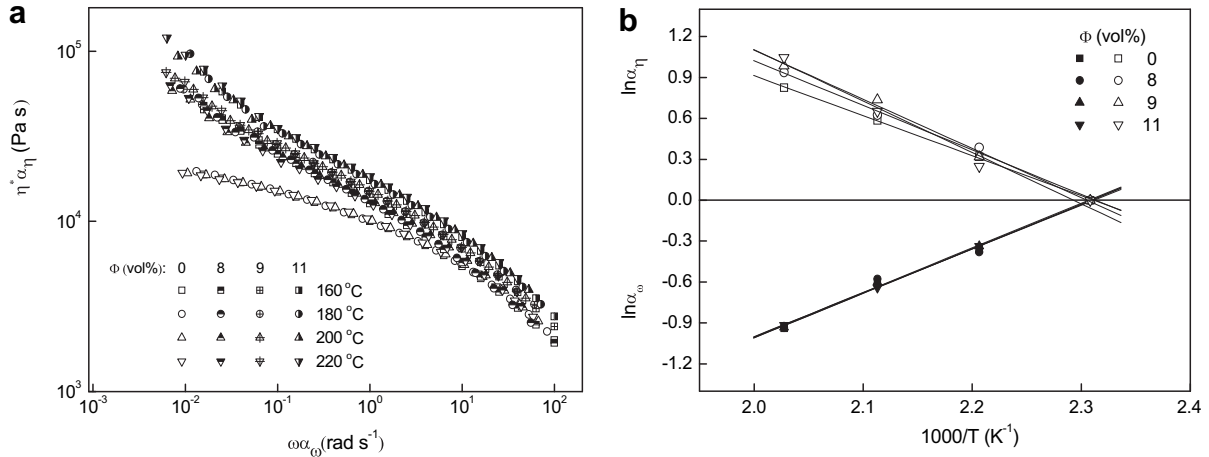


Fig. 4. Activation energies as a function of  $\Phi$ .

Fig. 6 gives  $G'$  and dynamic loss modulus ( $G''$ ) as a function of  $\omega$  at 180 °C for pure HDPE and the composites. At a given  $\omega$ , both  $G'$  and  $G''$  increase with increasing  $\Phi$  due to the reinforcement effect. Both  $G'$  and  $G''$  of the composites deviate from the classical linear relationships of  $G' \sim \omega^2$  and  $G'' \sim \omega$  at  $\omega \leq 1$ . In particular,  $G'$  and  $G''$  of the composite with  $\Phi = 11$  vol% appear second plateaus in terminal  $\omega$  zone, which was ascribed to the formation of three-dimensional CB network throughout the matrix [27,28].

Terminal relaxation time ( $\tau$ ) is an important parameter for evaluating polymer dynamics quantitatively [50].  $\tau$  was estimated as the reciprocal of characteristic frequency where  $G'$  and  $G''$  intersect with each other as shown in Fig. 6. Fig. 7 shows the Arrhenius plot of  $\tau$  against  $1/T$ . At given temperatures,  $\tau$  increases with increasing  $\Phi$ , reflecting the reduction of chain mobility due to the restriction of CB particles [44]. The  $\tau$  value is far smaller than  $t_{pR}$  and  $t_{pG}$ , indicating that the polymer dynamics may hardly influence the DC percolation and the DM percolation. The activation energy of terminal relaxation time,  $\Delta E_\tau$ , was determined from the slope of  $\tau$  against  $1/T$  and is plotted as a function of  $\Phi$  in Fig. 4.

As shown in Fig. 4,  $\Delta E_\eta$  increases slightly with increasing  $\Phi$  whereas the others are nearly independent of  $\Phi$ . In the earlier studies, Wu et al. investigated the DC percolation of CB-filled



**Fig. 5.** (a) Master curves of complex viscosity ( $\eta^*$ ) as a function of angular frequency ( $\omega$ ) at 160 °C for pure HDPE and for the composites measured at 160 °C, 180 °C, 200 °C, and 220 °C; (b) Arrhenius plot of the horizontal and vertical shift factors ( $\alpha_\eta$  and  $\alpha_\omega$ ) against  $1/T$ .

polymers and revealed that  $\Delta E_R$  is closely related to activation energy of zero-shear rate viscosity ( $\eta^0$ ) of the matrix [21]. Otherwise, Zhang et al. revealed that  $\Delta E_R$  approaches activation energy of  $\eta^0$  of filled polymer composites [49]. As shown in Fig. 4, both  $\Delta E_{pR}$  and  $\Delta E_{pG}$  are significantly higher than  $\Delta E_\eta$  and  $\Delta E_\omega$ , which are in disagreement with the speculations by Wu and Zhang. The results reveal that the dynamic percolation is more sensitive to temperature in comparison with viscoelasticity of the composites. The difference between  $\Delta E_C$  and  $\Delta E_R$  can be understood by taking the two percolation networks into consideration, i.e., the conduction network for charge carrier to transport and the combined network for mechanical momentum to transfer [38]. It is clear that both the DC percolation and the DM percolation are involved in the movement of CB aggregates. The percolation times are much longer than the terminal relaxation time of polymer chains.

### 3.4. Modeling the DC percolation

Classical percolation theory is well accepted for describing the conduction properties of conductive composites [51,52]. According to this theory,  $R$  as a function of  $\Phi$  can be expressed as

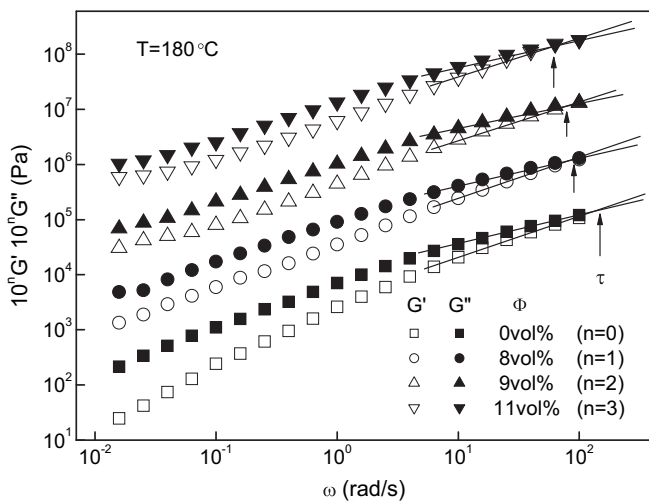
$$R = R_{0F} \left( \frac{\Phi - \Phi_C}{1 - \Phi_C} \right)^{-\beta} \quad (\Phi > \Phi_C) \quad (1)$$

$$R = R_{0M} \left( \frac{\Phi_C - \Phi}{\Phi_C} \right)^s \quad (\Phi < \Phi_C) \quad (2)$$

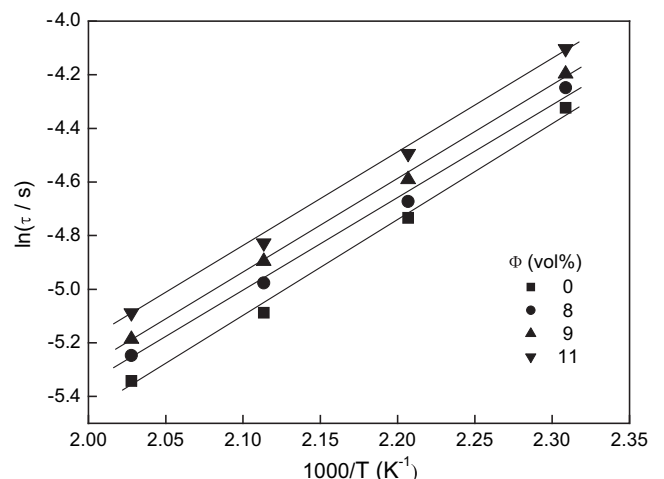
Here,  $\beta$  and  $s$  are exponents and  $\beta = 2$  is usually taken for three-dimensional systems [53–55].  $R_{0F}$  and  $R_{0M}$  are constants related to the filler and the matrix, respectively.

The conduction network preexisting at room temperature was partially broken down after the composites were heated to a temperature above  $T_m$ . During annealing above  $T_m$ , rearrangement of CB aggregates results in the reformation of continuous pathways responsible for the DC percolation. In this case,  $\Phi$  in Eqs. (1) and (2) should be substituted by a time-dependent effective volume fraction,  $\Phi_{\text{eff}}(t)$ , of interconnected CB aggregates.  $\Phi_C$  should also be replaced by another parameter  $\Phi_{\text{CR}}$  defined as the minimum CB volume fraction required for the occurrence of DC percolation. Otherwise, the composites with  $\Phi < \Phi_{\text{CR}}$  do not exhibit DC percolation. Under such treatment, Eqs. (1) and (2) can be modified as

$$R = R_{0F} \left( \frac{\Phi_{\text{eff}}(t) - \Phi_{\text{CR}}}{1 - \Phi_{\text{CR}}} \right)^{-\beta} \quad (\Phi > \Phi_C) \quad (3)$$



**Fig. 6.**  $G'$  and dynamic loss modulus ( $G''$ ) as a function of  $\omega$  at 180 °C for pure HDPE and the composites. The data of  $G'$  and  $G''$  are vertically shifted by  $10^n$  with the  $n$  values indicated.



**Fig. 7.** Relaxation time ( $\tau$ ) as a function of  $1/T$  for pure HDPE and the composites.

$$R = R_{OM} \left( \frac{\Phi_{CR} - \Phi_{eff}(t)}{\Phi_{CR}} \right)^s \quad (\Phi < \Phi_C) \quad (4)$$

Aggregation of colloidal systems can be described as a cluster–cluster aggregation process. The time-dependences of mean cluster size and its distribution are usually given by solving the Smoluchowski equation [56]. However, the mathematical solution of this equation is often very hard to get. A second order kinetics equation referring to the collision of two particles was proposed for describing the filler networking in composites [57]. The second order kinetics prevails for the particles homogeneously distributed in the matrix. However, for the partially broken down network studied here, the CB aggregates are inhomogeneously distributed so that it is hard for the process of “collision of two particles” to happen. On the contrary, the networking should follow the first order kinetics governed by the diffusion of isolated small aggregates towards immobilized large aggregates [39]. Neglecting difference in mobilities of aggregates with different sizes, the number of unconnected CB aggregates can reduce under the same kinetic constant  $K$ . The Smoluchowski rate equation can be simplified to the first order kinetics [14]

$$\frac{dN(t)}{dt} = KN(t) \quad (5)$$

Integrating Eq. (5) yields

$$\frac{N(t)}{N(\infty)} = 1 - \left[ 1 - \frac{N(0)}{N(\infty)} \right] \exp(-Kt) \quad (6)$$

or

$$P(t) = 1 - [1 - P(0)] \exp(-Kt) \quad (7)$$

Here,  $N(0)$ ,  $N(t)$ , and  $N(\infty)$  represents the number of interconnected CB particles per unit volume at time zero,  $t$  and infinite.  $P(0)$  and  $P(t)$  represent the fractions of interconnected CB aggregates at time  $t$  and infinite.

Assuming that all the CB aggregates are incorporated into the conduction network at time infinite, the time-dependent variable  $\Phi_{eff}(t)$  can be expressed as

$$\Phi_{eff}(t) = P(t)\Phi \quad (8)$$

In the DC percolation, continuous CB pathways begin to form only after  $t_{pR}$ . Then, the fraction of interconnected CB aggregates reaches  $\Phi_{CR}$  at  $t_{pR}$ ,

$$\Phi_{eff}(t_{pR}) = \Phi_{CR} = \Phi P(t_{pR}) = \Phi \{ 1 - [1 - P(0)] \exp(-Kt_{pR}) \} \quad (9)$$

Substituting Eqs. (7), (8) and (9) into Eqs. (3) and (4) yields

$$R = A [\exp(-Kt_{pR}) - \exp(-Kt)]^{-\beta} \quad (t > t_{pR}) \quad (10)$$

$$R = B [\exp(-Kt) - \exp(-Kt_{pR})]^s \quad (t < t_{pR}) \quad (11)$$

Here,  $A = R_{OF} \left\{ \frac{\Phi [1 - P(0)]}{1 - \Phi_{CR}} \right\}^{-\beta}$  and  $B = R_{OM} \left\{ \frac{\Phi [1 - P(0)]}{\Phi_{CR}} \right\}^s$  are constants depending on  $\Phi$  and  $T$ .

Fig. 8 presents  $\Phi$  against  $1/t_{pR}$  at different annealing temperatures.  $\Phi_{CR}$  is determined as 0.07 by linearly extrapolating  $1/t_{pR}$  to zero [49].  $\Phi_{CR}$  is irrespective of  $T$  and is the same as the percolation threshold at  $T > T_m$ , as shown in Fig. 1. Eqs. (10) and (11) are used to fit the experimental data of  $R$  against  $t$  above and below  $t_{pR}$ , respectively. The best fitting results are shown as dashed and dotted curves in Fig. 2. Table 1 lists the parameter values averaged

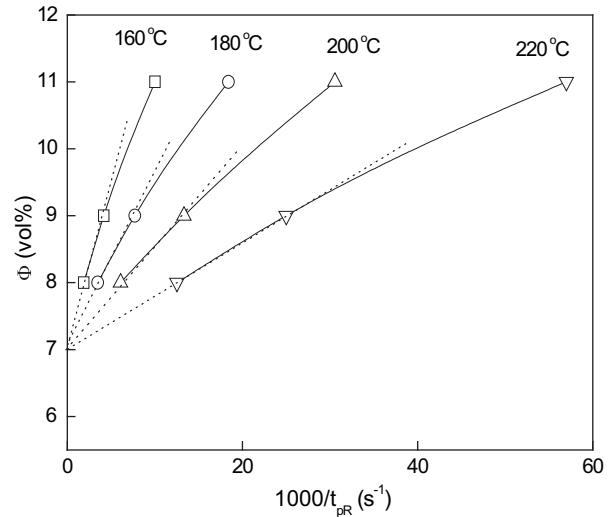


Fig. 8. Relationship between  $t_{pR}$  and  $\Phi$  at different annealing temperatures.

from at least three independent experiments. With the  $K$ ,  $t_{pR}$  and  $\Phi_{CR}$  values obtained, the value of  $1 - P(0)$  was calculated according to Eq. (9) and the results are listed in Table 1.

As shown in Table 1,  $K$  increases with increasing  $T$  but it is insensitive to  $\Phi$ , indicating that the movement of isolated CB aggregates is a diffusion-limited process [58]. Fig. 9 shows the relationship between  $K$  and  $1/\eta^0$  of HDPE. Here,  $\eta^0$  is taken as the  $\omega$ -independent  $\eta^*$  plateau value in the terminal region in Fig. 5a.  $K$  is linearly dependent on  $1/\eta^0$ , revealing that the diffusion of CB aggregates is closely related to the viscosity of the matrix. The parameters  $A$  and  $B$  decrease with increasing  $T$  and  $\Phi$ . The  $K$  values ( $2.0\text{--}7.0 \times 10^{-4} \text{ s}^{-1}$ ) are considerably small so that  $A$  approximates  $R$  at time infinite according to Eq. (10). The value of  $\beta$  decreases with increasing  $T$  and  $\Phi$ , showing the tendency to approach the universal value of 2. A variation of  $\beta$  value depending on CB grade as well as the type and microstructure of matrix has been observed in CB-filled rubbers at room temperature, which is ascribed to the kinetic aggregation of CB particles and the resultant alteration of the local structure of the percolation network [52]. The value of  $s$  is usually taken as 0.73 in classical percolation [59,60]. However,  $s$  is not a constant in the DC percolation, approaching to 0.73 with decreasing  $T$ .

The value of  $1 - P(0)$  decreases slightly with increasing  $T$ .  $P(0)$  is about 0.86, 0.77 and 0.62 for the composites with  $\Phi = 8 \text{ vol}\%$ ,  $9 \text{ vol}\%$  and  $11 \text{ vol}\%$ , suggesting that most CB aggregates are interconnected at the beginning of annealing.  $P(0)$  increases slightly with increasing  $T$ , resulting in a reduction of  $R$  with  $T$  at earlier stage of

Table 1  
Fitting parameters for the modified percolation model.

$\Phi$ (vol%)	$T$ (°C)	$K$ ( $10^{-4} \text{ s}^{-1}$ )	$A$ (k $\Omega$ )	$B$ ( $10^{10} \Omega$ )	$\beta$	$s$	$1 - P(0)$	$K_1$ ( $10^{-5} \text{ s}^{-1}$ )	$\alpha$
8	160	2.3	1500	11.0	3.40	0.78	0.143	2.3	0.10
	180	3.2	700	9.7	2.92	0.82	0.141	2.5	0.13
	200	4.5	300	9.1	2.75	0.85	0.138	2.3	0.15
	220	6.4	120	8.5	2.51	0.91	0.132	3.1	0.13
9	160	2.5	31	9.8	2.65	0.76	0.235	2.0	0.14
	180	3.3	12	6.9	2.52	0.85	0.231	2.7	0.17
	200	4.7	9.5	3.8	2.40	0.94	0.230	2.6	0.18
	220	6.3	6.5	2.5	2.22	1.06	0.228	2.5	0.15
11	160	2.5	20	8.2	2.51	0.74	0.375	2.4	0.15
	180	3.3	8.2	5.3	2.42	0.87	0.372	2.6	0.16
	200	4.6	6.5	2.7	2.35	1.07	0.367	2.6	0.16
	220	6.5	5.6	1.2	2.10	1.21	0.369	2.6	0.18

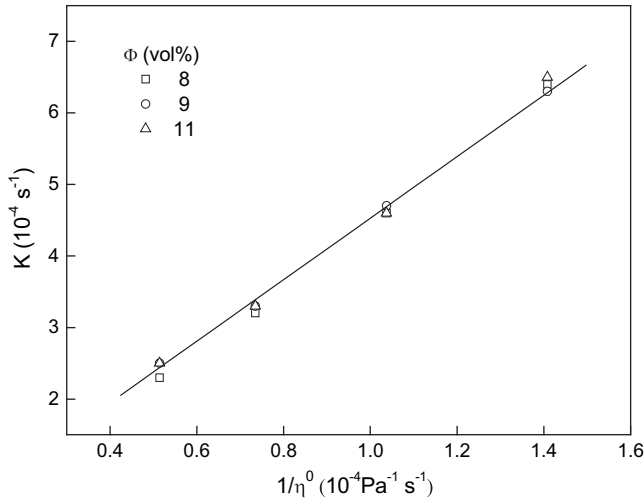


Fig. 9. Kinetic constant ( $K$ ) of the composites as a function of reciprocal zero-shear rate viscosity ( $1/\eta^0$ ) of the pure HDPE.

annealing as shown in Fig. 2. The variation of  $P(0)$  with respect to  $T$  could account for the negative temperature coefficient (NTC) effect of resistance usually observed in HDPE/CB composites above  $T_m$ . Upon heating the composites across  $T_m$  from below, the large volume expansion resulting from the matrix melting destroys the preexisting conduction network, leading to the partial breakdown of conduction network accompanied with the PTC effect. The DC percolation during annealing is not involved in all the aggregates with a wide distribution of size. Most likely, the diffusion of isolated small aggregates in the melt gradually interconnects the adjacent clusters, which rebuilds the conduction network upon annealing.

### 3.5. Modeling the DM percolation

As shown in Fig. 2, annealing causes  $G'$  to increase less than one order of magnitude at  $t \leq 10^4$  s. The variation amplitude is much smaller than the  $R$  decay at the same time scale. The difference in variation magnitudes of  $R$  and  $G'$  can be explained on the basis of different mechanisms. The former is related to the close CB electrical contact for charge carrier to transport, while the later is related to the viscous coupling between CB particles and polymer chains for mechanical momentum to transfer [39]. Being different from the DC percolation, polymer chains must participate in the DM percolation. There is no doubt that the chain bridging, together with the direct CB contact, plays an important role in the DM percolation.

$G'$  of filled polymers is usually expressed as a power law [48]

$$G' \sim (\Phi - \Phi_{CG})^\alpha (\Phi > \Phi_{CG}) \quad (12)$$

where  $\Phi_{CG}$  is a threshold of modulus and  $\alpha$  is a critical exponent. In the DM percolation,  $\Phi$  should be replaced by  $\Phi_{\text{eff}}(t)$ ,

$$G(t) \sim [P(t)\Phi - \Phi_{CG}]^\alpha (\Phi > \Phi_{CG}) \quad (13)$$

Being similar to the DC percolation, Eq. (13) can be modified to

$$G'(t) \sim [\Phi(1 - P(0))]^\alpha [\exp(-K_1 t_{pG}) - \exp(-K_1 t)]^\alpha (t > t_{pG}) \quad (14)$$

The kinetics parameter  $K_1$  differs from  $K$  because of the different mechanisms in the DM percolation and the DC percolation. Eq. (14) was applied to fit  $G'$  as a function of  $t$  using the least-square method and the fitting results are shown as solid curves in Fig. 2. The values

of  $K_1$  and  $\alpha$  averaged from at least three independent experiments are listed in Table 1. Both  $K_1$  and  $\alpha$  seem independent of  $T$  and  $\Phi$  and their average values are  $K_1 = (2.5 \pm 0.3) \times 10^{-5} \text{ s}^{-1}$  and  $\alpha = 0.15 \pm 0.02$ . The fitting results suggest that the DM percolation follows a universal law. The value of  $K_1$  is about one order of magnitude smaller than  $K$ , which is in agreement with the observation that the DM percolation is much slower than the DC percolation.

## 4. Conclusions

Thermal-induced DC percolation and DM percolation take place simultaneously during annealing HDPE/CB composites at  $T > T_m$ . The percolation times,  $t_{pR}$  and  $t_{pG}$ , are much longer than the terminal relaxation time of the composites. The activation energies,  $\Delta E_{pR}$  and  $\Delta E_{pG}$ , are much higher than those of viscous flow of the composites and of the terminal relaxation time of the matrix.

The first order kinetic particle aggregation model is introduced into the classical percolation theory to describe the DC percolation. The reformation of CB network, which has been partially broken down due to melting of the matrix, can well account for the DC percolation. Similar treatment is applicable to the DM percolation. The kinetics parameter  $K_1$  in the DM percolation is about one order of magnitude smaller than  $K$  in the DC percolation. The critical exponents  $\beta$  and  $s$  in the DC percolation are strongly dependent on  $T$  and  $\Phi$ . On the other hand, the critical exponent  $\alpha$  in the DM percolation seems to be a constant.

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